

## Chapter 12 Solutions Review

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Chapter 12 Solutions review. liquid that dissolves a gas or solid or the solution component in greater amount. when a system in equilibrium is altered by a change of temperature, pressure, or concentration variable, the system shifts in equilibrium composition in a way that tends to counteract this change of variable.

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compound is not an electrolyte, identify types of compounds, and calculate moles, grams, and molalities in various solutions.

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Chapter 12 Review: Solutions Name: Carly 1) Does a solution have to involve a liquid? Explain your answer. No; solutions may exist as gases, liquids, or solids. 2) Explain why a suspension is considered a heterogeneous mixture. The particles are so large that they settle out unless the mixture is constantly stirred or agitated.

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Modern Chemistry 5 Solutions CHAPTER 12 REVIEW Solutions SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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*Chapter 12 Review Solutions Section 3 Answers*

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CHAPTER 12 REVIEW Solutions SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the solvent in each of the following examples:

*mc06se cFMSr i-vi*

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*Chapter 12, Problem Review\_and\_Discussion 13 : 5. How does ...*

Modern Chemistry Chapter 12 Solutions Review Answers Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Chapter 12 Review Solutions Section 3 Answers

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