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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals. $\text{Br (l)} + 2\text{NaI (s)} \rightarrow 2\text{NaBr (s)} + \text{I}_2\text{(s)}$ 3. An aluminum strip is immersed in a solution of silver nitrate.

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(NH) CO ammonium oxalate 4 224 polonium(VI) thiocyanate Po(SCN)₆ tetraphosphorus decaoxide P₄O₁₀ 12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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